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Key indicators

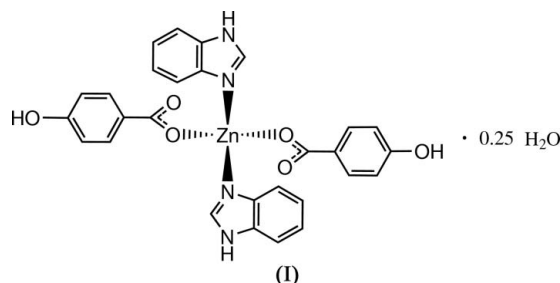
Single-crystal X-ray study
 $T = 295$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
Disorder in solvent or counterion
 R factor = 0.038
 wR factor = 0.098
Data-to-parameter ratio = 13.1For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Bis(1*H*-benzimidazole- κN^3)bis(4-hydroxy-
benzoato- κO)zinc(II) 0.25-hydrate

In the title compound, $[\text{Zn}(\text{C}_7\text{H}_5\text{O}_3)_2(\text{C}_7\text{H}_6\text{N}_2)_2] \cdot 0.25\text{H}_2\text{O}$, the Zn^{II} atom is coordinated by two 4-hydroxybenzoate anions and two benzimidazole (bzim) molecules, resulting in a distorted ZnO_2N_2 tetrahedral geometry. The short face-to-face distance of 3.28 (4) Å between parallel partially overlapped bzim ligands of neighboring molecules indicates the existence of π - π stacking in the crystal structure.

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Comment

As part of our ongoing investigation into the nature of π - π stacking in metal complexes, we have prepared a series of phases incorporating aromatic ligands such as quinoline (Pan & Xu, 2004) and benzimidazole (bzim) (Li *et al.*, 2005; Zheng *et al.*, 2005). We present here the title mixed-ligand Zn^{II} complex, (I) (Fig. 1), for comparison with the structures reported previously.



Compound (I) is isostructural with a Co^{II} complex (Zheng *et al.*, 2005), in which a compressed $\text{Co}-\text{O}-\text{C}$ angle [103.5 (2)°] suggested a long but genuine $\text{Co}-\text{O}$ bonding interaction of 2.561 (3) Å (*i.e.* a highly asymmetric bidentate coordination by the carboxylate group of the 4-hydroxybenzoate anion in question). As shown in Table 1, the $\text{Zn}-\text{O}$ and $\text{Zn}-\text{N}$ bond distances in (I) are comparable with the corresponding normal $\text{Co}-\text{O}$ bond distances [1.970 (2) and 1.993 (2) Å] and $\text{Co}-\text{N}$ bond distances [2.026 (3) and 2.013 (3) Å], but the $\text{Zn} \cdots \text{O5}$ separation of 2.693 (2) Å is significantly longer than the corresponding $\text{Co}-\text{O}$ bond in the Co^{II} complex. The $\text{Zn}-\text{O4}-\text{C41}$ angle of 108.2 (2)° is also larger than the compressed $\text{Co}-\text{O}-\text{C}$ angle.

These findings suggest there is no significant bonding interaction between atoms Zn and O5 in (I). Thus, the metal coordination polyhedron for (I) is a somewhat distorted ZnO_2N_2 tetrahedron, compared to distorted trigonal bipyramidal for the Co^{II} complex. The difference in coordination number and coordination geometry between the two compounds might arise from the different electronic configurations of the two metal ions.

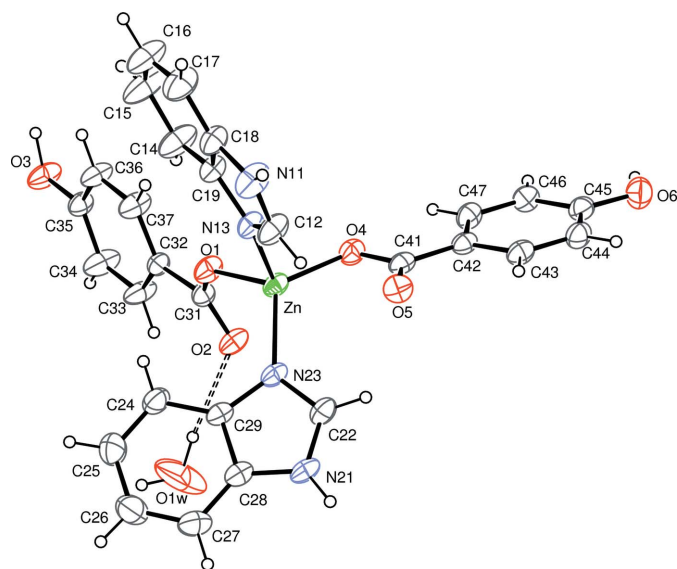


Figure 1
The molecular structure of (I), shown with 30% probability displacement ellipsoids (arbitrary spheres for H atoms). The dashed line indicates a hydrogen bond.

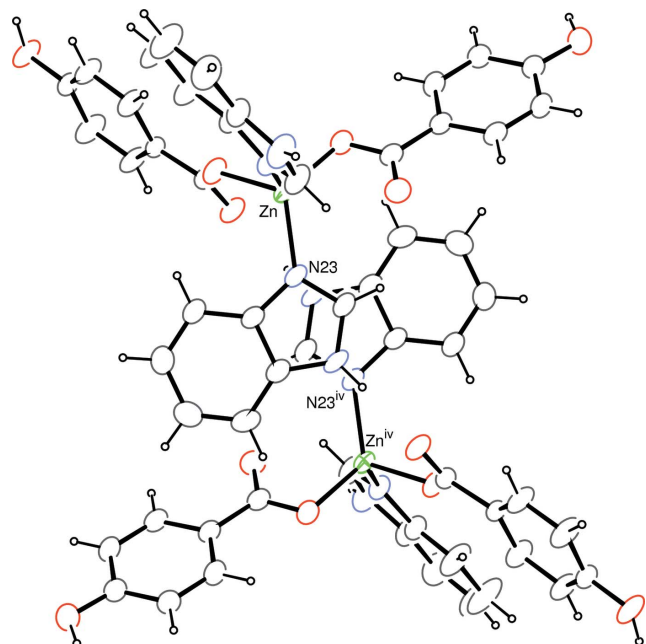


Figure 2
The π - π stacking between parallel N23-imidazole and N23^{iv}-imidazole rings in (I) [symmetry code: (iv) $\frac{3}{2} - x, \frac{3}{2} - y, 1 - z$].

A partially overlapped arrangement is observed between parallel N23-imidazole and N23^{iv}-imidazole rings (see Fig. 2 for symmetry code). The face-to-face separation of 3.28 (4) Å indicates the existence of π - π stacking. The disordered non-coordinated water molecule, situated close to a twofold axis, is hydrogen bonded to the uncoordinated carboxyl O2 atom. Further O-H...O and N-H...O interactions (Table 2) involving the ligands help to stabilize the crystal structure of (I).

Experimental

Commercial analytical grade reagents were used. ZnCl₂ (1 mmol), 4-hydroxybenzoic acid (2 mmol), bzim (2 mmol) and Na₂CO₃ (1 mmol) were dissolved in a mixture of water (15 ml) and ethanol (5 ml). The solution was refluxed for 4 h and filtered after cooling to room temperature. Single crystals of (I) were obtained from the filtrate after 3 d.

Crystal data

[Zn(C ₇ H ₅ O ₃) ₂ (C ₇ H ₆ N ₂) ₂].0.25H ₂ O	$D_x = 1.479 \text{ Mg m}^{-3}$
$M_r = 580.37$	Mo K α radiation
Monoclinic, C2/c	Cell parameters from 5766 reflections
$a = 13.064 (3) \text{ \AA}$	$\theta = 3.0\text{--}23.0^\circ$
$b = 27.434 (6) \text{ \AA}$	$\mu = 0.99 \text{ mm}^{-1}$
$c = 15.069 (4) \text{ \AA}$	$T = 295 (2) \text{ K}$
$\beta = 105.138 (4)^\circ$	Block, colorless
$V = 5213 (2) \text{ \AA}^3$	$0.29 \times 0.20 \times 0.11 \text{ mm}$
$Z = 8$	

Data collection

Rigaku R-Axis RAPID diffractometer	3148 reflections with $I > 2\sigma(I)$
ω scans	$R_{\text{int}} = 0.035$
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)	$\theta_{\text{max}} = 25.2^\circ$
$T_{\text{min}} = 0.745, T_{\text{max}} = 0.902$	$h = -15 \rightarrow 15$
13925 measured reflections	$k = -32 \rightarrow 25$
4714 independent reflections	$l = -16 \rightarrow 18$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0379P)^2 + 5.4788P]$
$R[F^2 > 2\sigma(F^2)] = 0.038$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.098$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.02$	$\Delta\rho_{\text{max}} = 0.34 \text{ e \AA}^{-3}$
4714 reflections	$\Delta\rho_{\text{min}} = -0.24 \text{ e \AA}^{-3}$
361 parameters	
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

Zn—O1	1.957 (2)	Zn—N13	2.010 (2)
Zn—O4	1.982 (2)	Zn—N23	1.991 (3)
O1—Zn—O4	102.03 (10)	O4—Zn—N13	110.52 (10)
O1—Zn—N13	102.08 (10)	O4—Zn—N23	120.39 (10)
O1—Zn—N23	112.11 (10)	N13—Zn—N23	108.15 (11)

Table 2

Hydrogen-bond geometry (Å, °).

D—H...A	D—H	H...A	D...A	D—H...A
O1W—H1A...O2	0.97	2.39	3.358 (13)	174
O3—H3A...O4 ⁱ	0.98	1.77	2.701 (3)	157
O6—H6A...O2 ⁱⁱ	0.95	1.75	2.698 (4)	174
N11—H11...O3 ⁱⁱⁱ	0.86	2.04	2.895 (4)	174
N21—H21...O2 ^{iv}	0.86	1.95	2.804 (4)	172

Symmetry codes: (i) $-x + 1, -y + 1, -z + 1$; (ii) $-x + 1, y, -z + \frac{3}{2}$; (iii) $x + 1, -y + 1, z + \frac{1}{2}$; (iv) $-x + \frac{3}{2}, -y + \frac{3}{2}, -z + 1$.

The non-coordinated water molecule is situated close to a twofold axis such that O1W...O1Wⁱ [symmetry code: (i) $1 - x, y, \frac{1}{2} - z$] = 1.79 Å. Its site-occupancy factor was refined and converged to

0.258 (9), and was fixed as 0.25 in the final cycles of refinement; this water molecule is thus disordered. The H atoms attached to O1W were placed in theoretical positions (Nardelli, 1999). The H atoms of the hydroxy groups were located in a difference map. The O-bound H atoms were refined as riding in their as-found relative positions, with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$. Other H atoms were placed in calculated positions, with C—H = 0.93 Å and N—H = 0.86 Å, and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier})$.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2002); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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